



1
00:00:11,419 --> 00:00:09,589
and so I gave this talk last year but

2
00:00:14,270 --> 00:00:11,429
now with a thousand percent more results

3
00:00:16,310 --> 00:00:14,280
in conclusions so yes this is the

4
00:00:18,380 --> 00:00:16,320
potential photochemical origins of

5
00:00:22,310 --> 00:00:18,390
banded iron formations I'm Parker

6
00:00:24,980 --> 00:00:22,320
casselberry at ASU so first banded iron

7
00:00:27,470 --> 00:00:24,990
formations there are large sedimentary

8
00:00:32,359 --> 00:00:27,480
deposits of iron oxide by large I mean

9
00:00:34,760 --> 00:00:32,369
trillions of tons these are laser these

10
00:00:37,190 --> 00:00:34,770
are very pretty ones in Western

11
00:00:41,350 --> 00:00:37,200
Australia and they're alternating layers

12
00:00:44,720 --> 00:00:41,360
of high iron and low iron minerals and

13
00:00:46,880 --> 00:00:44,730

they're thought to be formed by oxygen

14

00:00:49,639 --> 00:00:46,890

oxidizing iron in seawater that

15

00:00:56,209 --> 00:00:49,649

precipitates and the thought for this

16

00:00:58,639 --> 00:00:56,219

comes from if you look at oxygen levels

17

00:01:01,040 --> 00:00:58,649

in the atmosphere over time and biff

18

00:01:03,470 --> 00:01:01,050

relative abundance they correlate well

19

00:01:06,859 --> 00:01:03,480

there's a lot of this deposited right

20

00:01:09,280 --> 00:01:06,869

around the great oxidation event but

21

00:01:12,380 --> 00:01:09,290

there are some alternatives proposed

22

00:01:15,380 --> 00:01:12,390

there's photo Pharaoh trophy which is a

23

00:01:17,899 --> 00:01:15,390

biological process it's type of an

24

00:01:21,020 --> 00:01:17,909

oxygen acronis asst that directly

25

00:01:22,880 --> 00:01:21,030

oxidizes iron and experiments and

26

00:01:25,190 --> 00:01:22,890

modeling have shown that you could

27

00:01:28,190 --> 00:01:25,200

generate enough oxidized iron to make

28

00:01:31,179 --> 00:01:28,200

Biff's with this and that is a beautiful

29

00:01:34,429 --> 00:01:31,189

lake where photo photography is studied

30

00:01:36,590 --> 00:01:34,439

but there's another idea that is

31

00:01:40,280 --> 00:01:36,600

photooxidation this is a completely

32

00:01:42,679 --> 00:01:40,290

abiotic process and it's ultraviolet

33

00:01:45,499 --> 00:01:42,689

light turning iron 2 into iron 3 and

34

00:01:49,700 --> 00:01:45,509

this is a cartoon I have somewhat

35

00:01:51,859 --> 00:01:49,710

abstracted mechanistic details so you

36

00:01:54,190 --> 00:01:51,869

have an ultraviolet photon from the Sun

37

00:01:59,090 --> 00:01:54,200

hits an iron two species in sea water

38

00:02:03,560 --> 00:01:59,100

turns it into iron hydroxide and then

39

00:02:06,170 --> 00:02:03,570

that will precipitate soluble and then

40

00:02:09,729 --> 00:02:06,180

over time that is transformed into Biff

41

00:02:12,890 --> 00:02:09,739

minerals like hematite and magnetite and

42

00:02:13,790 --> 00:02:12,900

the implication here is that as I said

43

00:02:16,850 --> 00:02:13,800

this is an a

44

00:02:20,360 --> 00:02:16,860

biotic process so you don't need oxygen

45

00:02:23,720 --> 00:02:20,370

or even biological activity so are these

46

00:02:27,740 --> 00:02:23,730

bits that we're seeing really signs of

47

00:02:30,400 --> 00:02:27,750

oxygen in our life or neither so

48

00:02:33,860 --> 00:02:30,410

previous work has been done testing this

49

00:02:36,650 --> 00:02:33,870

in 1983 brighter min at all used a

50

00:02:39,770 --> 00:02:36,660

mercury lamp and they observed reactions

51
00:02:44,570 --> 00:02:39,780
using filters for long wavelengths over

52
00:02:47,420 --> 00:02:44,580
400 nanometers that's in the visible not

53
00:02:49,790 --> 00:02:47,430
even in the UV so that led them to

54
00:02:51,860 --> 00:02:49,800
believe that this iron species Fe^{2+}

55
00:02:54,770 --> 00:02:51,870
was responsible because it has

56
00:02:59,170 --> 00:02:54,780
absorbance going out to almost 450

57
00:03:02,710 --> 00:02:59,180
nanometers and this is important because

58
00:03:07,340 --> 00:03:02,720
so this is the solar radiation spectrum

59
00:03:11,600 --> 00:03:07,350
and here's the UV so the earth had no

60
00:03:13,880 --> 00:03:11,610
ozone back then so it was somewhat

61
00:03:18,410 --> 00:03:13,890
proximate to this top of atmosphere

62
00:03:22,010 --> 00:03:18,420
spectrum there and so if you have Fe^{2+}

63
00:03:23,900 --> 00:03:22,020

it absorbs a huge region of UV and

64

00:03:26,449 --> 00:03:23,910

getting into the visible where there's a

65

00:03:28,070 --> 00:03:26,459

lot more radiation as one of my

66

00:03:32,930 --> 00:03:28,080

professor said that's the meaty part of

67

00:03:35,930 --> 00:03:32,940

the solar spectrum and Francois took

68

00:03:37,580 --> 00:03:35,940

these results ran with them and made

69

00:03:42,110 --> 00:03:37,590

notion model calculated a mass

70

00:03:44,479 --> 00:03:42,120

cumulation rate and he got about an

71

00:03:48,590 --> 00:03:44,489

order of magnitude more than an

72

00:03:52,600 --> 00:03:48,600

estimated natural rate for a bit so that

73

00:03:57,740 --> 00:03:52,610

led people to conclude this is plausible

74

00:04:00,699 --> 00:03:57,750

but I went back to re-examine this so I

75

00:04:03,790 --> 00:04:00,709

used a quartz reactor with ports for

76
00:04:06,050 --> 00:04:03,800
flowing gas inventing drawing samples

77
00:04:10,009 --> 00:04:06,060
continuously purged with nitrogen co2

78
00:04:13,310 --> 00:04:10,019
gas which 2% co2 is not unreasonable for

79
00:04:16,250 --> 00:04:13,320
the archaean and that's to keep oxygen

80
00:04:19,880 --> 00:04:16,260
out because this is extremely sensitive

81
00:04:22,880 --> 00:04:19,890
doxygen solution was deoxygenated di

82
00:04:27,230 --> 00:04:22,890
water sodium chloride iron and

83
00:04:31,100 --> 00:04:27,240
bicarbonate so this is what I saw

84
00:04:35,390 --> 00:04:31,110
24 hours and in the bottom there are

85
00:04:37,400 --> 00:04:35,400
those kind of orange precipitates Raman

86
00:04:39,499 --> 00:04:37,410
spectroscopy identified those as lipid

87
00:04:44,960 --> 00:04:39,509
occurro site which is consistent with the

88
00:04:46,760 --> 00:04:44,970

earlier work so then though well the

89

00:04:52,939 --> 00:04:46,770

mercury lamp poorly matches the solar

90

00:04:55,219 --> 00:04:52,949

spectrum it has all these peaks and so I

91

00:04:57,890 --> 00:04:55,229

used a solar simulator as a light source

92

00:05:01,189 --> 00:04:57,900

which is a Zeon arc lamp and special

93

00:05:04,999 --> 00:05:01,199

filters to match the solar spectrum but

94

00:05:09,830 --> 00:05:05,009

if you FeO H⁺ is absorbing then I should

95

00:05:13,730 --> 00:05:09,840

observe a reaction so this is the dark

96

00:05:16,510 --> 00:05:13,740

control no precipitation this iron was

97

00:05:19,219 --> 00:05:16,520

measured on quadrupole ICP-MS and

98

00:05:21,230 --> 00:05:19,229

everything was ratio to sodium to

99

00:05:25,430 --> 00:05:21,240

correct for any evaporation because

100

00:05:28,370 --> 00:05:25,440

these experiments took days and dramatic

101
00:05:34,580 --> 00:05:28,380
pause solar simulator also showed

102
00:05:41,390 --> 00:05:34,590
nothing no reaction over 46 hours this

103
00:05:43,939 --> 00:05:41,400
was quite puzzling but the mercury lamp

104
00:05:48,560 --> 00:05:43,949
and the solar simulator have extremely

105
00:05:54,409 --> 00:05:48,570
different spectra so there's just a few

106
00:05:57,290 --> 00:05:54,419
slight differences so what I did is to

107
00:05:59,120 --> 00:05:57,300
test the wavelength dependence you know

108
00:06:02,209 --> 00:05:59,130
which of these differences is important

109
00:06:04,879 --> 00:06:02,219
I performed an experiment using the

110
00:06:08,469 --> 00:06:04,889
mercury lamp and light filters it was

111
00:06:10,879 --> 00:06:08,479
one continuous experiment and

112
00:06:14,629 --> 00:06:10,889
periodically swapping out light filters

113
00:06:16,909 --> 00:06:14,639

and so if I change light filters and saw

114

00:06:23,209 --> 00:06:16,919

a reaction oh that's the wavelength that

115

00:06:26,390 --> 00:06:23,219

caused it so to begin I started with no

116

00:06:31,850 --> 00:06:26,400

light dark control and so I know our

117

00:06:34,370 --> 00:06:31,860

precipitation okay good and then light

118

00:06:37,760 --> 00:06:34,380

longer than 400 nanometers also showed

119

00:06:39,629 --> 00:06:37,770

no effect this is different than what

120

00:06:42,809 --> 00:06:39,639

Khan house are observed or

121

00:06:44,610 --> 00:06:42,819

not seem brighter Minh there's another

122

00:06:48,689 --> 00:06:44,620

paper con Howser that does agree with

123

00:06:51,420 --> 00:06:48,699

this and then light longer than three

124

00:06:53,519 --> 00:06:51,430

hundred and forty five nanometers also

125

00:07:00,269 --> 00:06:53,529

showed no effect so that is getting down

126
00:07:02,309 --> 00:07:00,279
into the UV and then boom 295 2 340 5

127
00:07:06,469 --> 00:07:02,319
nanometers cause significant iron

128
00:07:09,510 --> 00:07:06,479
precipitation in a couple days and then

129
00:07:14,339 --> 00:07:09,520
the full spectrum didn't actually show

130
00:07:16,409 --> 00:07:14,349
an increased rate so I can conclude oh

131
00:07:20,939 --> 00:07:16,419
and precipitation stopped when the light

132
00:07:25,860 --> 00:07:20,949
was turned off just to make sure so I

133
00:07:28,409 --> 00:07:25,870
can conclude that mid UV so 345

134
00:07:35,579 --> 00:07:28,419
nanometers or so is what's causing this

135
00:07:37,439 --> 00:07:35,589
reaction and then so what I did is using

136
00:07:40,950 --> 00:07:37,449
this new wavelength dependence that I

137
00:07:44,010 --> 00:07:40,960
found I went back to the old models

138
00:07:46,469 --> 00:07:44,020

where you calculated it using only that

139

00:07:50,279 --> 00:07:46,479

region that I observed and not the

140

00:07:54,390 --> 00:07:50,289

long-wavelength where you have all of

141

00:07:59,189 --> 00:07:54,400

the higher solar radiation and I

142

00:08:01,709 --> 00:07:59,199

calculated rates that are 6 250 4

143

00:08:03,570 --> 00:08:01,719

milligrams of iron per square centimeter

144

00:08:06,540 --> 00:08:03,580

per year depending on the concentration

145

00:08:08,129 --> 00:08:06,550

of iron too there's a lot of

146

00:08:10,679 --> 00:08:08,139

uncertainties and assumptions that went

147

00:08:14,279 --> 00:08:10,689

into this but that's just to get kind of

148

00:08:17,010 --> 00:08:14,289

a rough estimate and like i said i'm

149

00:08:21,059 --> 00:08:17,020

missing the longer wave absorbance here

150

00:08:23,999 --> 00:08:21,069

so this is five times less than previous

151
00:08:28,980 --> 00:08:24,009
models five times slower but it actually

152
00:08:34,110 --> 00:08:28,990
still compares favorably to estimated

153
00:08:36,060 --> 00:08:34,120
natural rates but looking at I talked

154
00:08:39,870 --> 00:08:36,070
about those alternative theories like

155
00:08:42,269 --> 00:08:39,880
photo Farrah trophy foto para trophy

156
00:08:45,930 --> 00:08:42,279
there's actually you can get numbers of

157
00:08:50,120 --> 00:08:45,940
around 500 milligrams per centimeter per

158
00:08:54,020 --> 00:08:50,130
year for similar iron concentrations

159
00:08:58,480 --> 00:08:54,030
so I can conclude here that I

160
00:09:01,220 --> 00:08:58,490
reevaluated iron photooxidation and

161
00:09:05,120 --> 00:09:01,230
found it sensitive to wavelengths

162
00:09:09,290 --> 00:09:05,130
between 295 and 345 nanometers it

163
00:09:13,610 --> 00:09:09,300

results in photooxidation rates 56 times

164

00:09:16,250 --> 00:09:13,620

slower for a predicted model and so if

165

00:09:19,190 --> 00:09:16,260

photo Farrah trophy was around or like

166

00:09:21,920 --> 00:09:19,200

oxygen for moxa net photosynthesis that

167

00:09:25,370 --> 00:09:21,930

would likely outpace this process but

168

00:09:32,840 --> 00:09:25,380

this still is possible in the absence of

169

00:09:35,030 --> 00:09:32,850

any biology at all so questions actually

170

00:09:39,440 --> 00:09:35,040

I have some time so I can talk a little

171

00:09:42,440 --> 00:09:39,450

bit about future work I said I had

172

00:09:44,420 --> 00:09:42,450

dissolved carbonate in there another

173

00:09:47,300 --> 00:09:44,430

thing I've been wanting to test

174

00:09:50,060 --> 00:09:47,310

different additives to this thing's have

175

00:09:52,270 --> 00:09:50,070

redox interactions with the iron also

176
00:09:54,740 --> 00:09:52,280
the effects of dissolved organic carbon

177
00:09:56,960 --> 00:09:54,750
because I could see a photo reduction

178
00:10:00,170 --> 00:09:56,970
which is what happens a lot in modern

179
00:10:01,760 --> 00:10:00,180
systems so I might actually get some

180
00:10:07,340 --> 00:10:01,770
filtered sea water and do experiments

181
00:10:09,140 --> 00:10:07,350
with that and then after this there's

182
00:10:19,630 --> 00:10:09,150
other things I could look at

183
00:10:27,340 --> 00:10:22,390
and I do have time for questions okay

184
00:10:29,380 --> 00:10:27,350
didn't even have to ask so I had a

185
00:10:33,640 --> 00:10:29,390
question about your relative xenon arc

186
00:10:35,290 --> 00:10:33,650
lamp and it's mercury so um I mean xenon

187
00:10:37,690 --> 00:10:35,300
arc lamps have a lot of output in that

188
00:10:40,480 --> 00:10:37,700

same near UV range that you're you're

189

00:10:42,610 --> 00:10:40,490

looking at so what is the like wattage

190

00:10:44,500 --> 00:10:42,620

of your various lamps like is it just

191

00:10:45,910 --> 00:10:44,510

coming from a difference there cuz if

192

00:10:53,220 --> 00:10:45,920

you look at the spectra using it on our

193

00:11:01,270 --> 00:10:58,120

referring to here yeah yeah um the

194

00:11:06,490 --> 00:11:01,280

wattage it was 300 watts okay I mean mu

195

00:11:09,580 --> 00:11:06,500

I guess my question is um the xenon arc

196

00:11:11,350 --> 00:11:09,590

lamps you know have a lot of energy in

197

00:11:13,510 --> 00:11:11,360

that range too so maybe it's just a

198

00:11:18,220 --> 00:11:13,520

power issue with why you weren't seeing

199

00:11:22,600 --> 00:11:18,230

it it's a power issue um I also I have

200

00:11:27,220 --> 00:11:22,610

ideas the air mass filter that I put on

201
00:11:29,110 --> 00:11:27,230
this because that corrects for more of

202
00:11:33,270 --> 00:11:29,120
the longer wavelengths that I was

203
00:11:39,880 --> 00:11:37,780
yeah is actually having some absorbance

204
00:11:44,920 --> 00:11:39,890
right in this region cutting down on the

205
00:11:47,170 --> 00:11:44,930
xenon arc spectrum so essentially the

206
00:11:49,180 --> 00:11:47,180
air mass filter for a solar simulator

207
00:11:53,980 --> 00:11:49,190
might not actually be the best solar

208
00:11:56,110 --> 00:11:53,990
simulator for these UV wavelengths but

209
00:11:57,790 --> 00:11:56,120
yeah I'm thinking about doing that also

210
00:12:03,040 --> 00:11:57,800
talked about getting a deuterium arc

211
00:12:08,490 --> 00:12:03,050
lamp for a I guess consistent UV

212
00:12:11,210 --> 00:12:08,500
spectrum longer wave or shorter wave but

213
00:12:15,330 --> 00:12:11,220

yes

214

00:12:18,000 --> 00:12:15,340

I was wondering so the solar spectrum

215

00:12:21,660 --> 00:12:18,010

that you're showing here is this for

216

00:12:23,340 --> 00:12:21,670

acquiescence Sun or like a flaring Sun

217

00:12:24,690 --> 00:12:23,350

or this is my solar simulator right

218

00:12:28,980 --> 00:12:24,700

right but it's supposed to simulate

219

00:12:31,800 --> 00:12:28,990

aqueous and yes fun okay i guess i'm

220

00:12:37,110 --> 00:12:31,810

just wondering since you know the Sun

221

00:12:46,800 --> 00:12:37,120

does his evolution that is one of my

222

00:12:50,390 --> 00:12:46,810

favorite questions there we go so here

223

00:12:54,180 --> 00:12:50,400

is the evolution of the solar spectrum

224

00:12:56,700 --> 00:12:54,190

and while the solar simulator didn't

225

00:13:00,810 --> 00:12:56,710

take this into account I actually took

226

00:13:04,590 --> 00:13:00,820

this into account in my models but the

227

00:13:08,430 --> 00:13:04,600

Sun was in 20 25 30 percent fainter

228

00:13:11,070 --> 00:13:08,440

early in its history so this is photon

229

00:13:13,110 --> 00:13:11,080

flux and the black is the modern

230

00:13:16,830 --> 00:13:13,120

spectrum and then these are different

231

00:13:18,720 --> 00:13:16,840

ages for the Sun so 100 million years

232

00:13:23,550 --> 00:13:18,730

700 million years and 2 billion years

233

00:13:25,470 --> 00:13:23,560

old and the interesting thing is that

234

00:13:28,980 --> 00:13:25,480

because I heard a lot of theories well

235

00:13:32,370 --> 00:13:28,990

the Sun was brighter in the UV but

236

00:13:34,950 --> 00:13:32,380

actually that is true but only for

237

00:13:39,240 --> 00:13:34,960

wavelengths under 200 nanometers which

238

00:13:41,480 --> 00:13:39,250

aren't of interest for this reaction so

239

00:13:45,870 --> 00:13:41,490

they're generated by different processes

240

00:13:50,730 --> 00:13:45,880

so does that answer question i think i'm

241

00:13:52,410 --> 00:13:50,740

referring more to the solar cycle so oh

242

00:14:00,430 --> 00:13:52,420

I thinks i see i see how you get on

243

00:14:08,030 --> 00:14:06,440

more questions so given the timing of

244

00:14:10,550 --> 00:14:08,040

biff deposits that we see in the rock

245

00:14:13,070 --> 00:14:10,560

record how much of a contribution do you

246

00:14:14,630 --> 00:14:13,080

think this photo-oxidation had on those

247

00:14:24,350 --> 00:14:14,640

formation of you know real world

248

00:14:30,970 --> 00:14:24,360

deposits right actually as i scroll back

249

00:14:34,190 --> 00:14:30,980

through all my slides yeah so I mean

250

00:14:41,140 --> 00:14:34,200

what i think is entirely possible is

251

00:14:44,030 --> 00:14:41,150

that things like older Biff's here were

252

00:14:47,810 --> 00:14:44,040

deposited by photochemistry and then if

253

00:14:50,720 --> 00:14:47,820

at some point photo farah trophy evolved

254

00:14:54,110 --> 00:14:50,730

or accidental photosynthesis then that

255

00:14:55,850 --> 00:14:54,120

is what caused this huge increase so

256

00:14:58,580 --> 00:14:55,860

there might be some background of

257

00:15:11,570 --> 00:14:58,590

photochemistry and then on top of that a

258

00:15:13,400 --> 00:15:11,580

large biogenic signal okay i am not at

259

00:15:14,990 --> 00:15:13,410

all very familiar with photo chemistry

260

00:15:18,140 --> 00:15:15,000

so I have a very basic question for you

261

00:15:20,840 --> 00:15:18,150

um in foot in the photooxidation of the

262

00:15:24,470 --> 00:15:20,850

reduced iron where is the oxygen coming

263

00:15:27,080 --> 00:15:24,480

from the oxen oh so you're saying is it

264

00:15:29,570 --> 00:15:27,090

like dissolved carbonate that oxygen

265

00:15:33,890 --> 00:15:29,580

yeah that oxygen come from water

266

00:15:35,900 --> 00:15:33,900

actually the oxidized iron takes a

267

00:15:49,770 --> 00:15:35,910

nearby water molecule and is like gimme

268

00:15:57,190 --> 00:15:55,930

so since the UM oh um the OA chums from

269

00:16:02,020 --> 00:15:57,200

the water don't you get a build up of

270

00:16:05,530 --> 00:16:02,030

hydrogen gas that's reducing well you

271

00:16:08,500 --> 00:16:05,540

get hydrogen gas yes but actually this

272

00:16:10,600 --> 00:16:08,510

is in the ocean and it actually escaped

273

00:16:13,870 --> 00:16:10,610

and goes into the app so in your setup

274

00:16:15,880 --> 00:16:13,880

it's not an ocean it's smaller so what

275

00:16:17,890 --> 00:16:15,890

what does what do you do with the

276

00:16:19,960 --> 00:16:17,900

hydrogen nuts well I'm continuously

277

00:16:22,390 --> 00:16:19,970

flowing gas through my experiment oh so

278

00:16:26,170 --> 00:16:22,400

there's an in and out yes oh um in fact

279

00:16:28,690 --> 00:16:26,180

I didn't talk about it but I didn't talk

280

00:16:31,630 --> 00:16:28,700

about a gas flow rate but i'm actually

281

00:16:34,870 --> 00:16:31,640

at 500 milliliters per minute I have a

282

00:16:37,300 --> 00:16:34,880

huge doer of liquid nitrogen to keep

283

00:16:39,640 --> 00:16:37,310

this experiment going okay I have tried

284

00:16:43,300 --> 00:16:39,650

lower flow rates but oxygen actually

285

00:16:46,720 --> 00:16:43,310

defuses in because a few i calculated a

286

00:16:49,470 --> 00:16:46,730

few ppm oxygen will give me oxidation

287

00:16:55,540 --> 00:16:49,480

rates that I observed with the lights

288

00:16:59,520 --> 00:16:55,550

okay so um just maybe this may be a

289

00:17:01,960 --> 00:16:59,530

mother geological question along it so

290

00:17:03,730 --> 00:17:01,970

due to the fact I guess there's certain

291

00:17:05,410 --> 00:17:03,740

period of time where parts of the rocky

292

00:17:06,760 --> 00:17:05,420

mountains were actually underwater so is

293

00:17:10,150 --> 00:17:06,770

this actually how they're getting the

294

00:17:14,520 --> 00:17:10,160

the the bands on certain rock formations

295

00:17:20,440 --> 00:17:14,530

that are actually are on land today hmm

296

00:17:31,280 --> 00:17:23,990

but I mean yeah I'm not sure about